Flame Plasma Surface Treating – What’s it All About

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Introduction

Flame treaters are used in two basic applications, Pre-Treating and Post Treating.

Pre-Treating is defined as the application of flame to paper board prior to extrusion coating.

Post Treatment is defined as the application of a “controlled” flame to the PE coated paper board or directly to a film substrate.

Examples of some of the types of substrates flame treated arte shown in Table 1.

Table 1. Substrates Processed

<table>
<thead>
<tr>
<th>SUBSTRATES PROCESSED</th>
<th>COATING</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>POLYMERS</td>
</tr>
<tr>
<td>Polyolefin</td>
<td>Primer topcoat</td>
</tr>
<tr>
<td>BOPP</td>
<td>“</td>
</tr>
<tr>
<td>BOPP</td>
<td>“</td>
</tr>
<tr>
<td>PET</td>
<td>“</td>
</tr>
<tr>
<td></td>
<td>PAPER S</td>
</tr>
<tr>
<td>Clay coated Kraft</td>
<td>Silicone Topcoat</td>
</tr>
<tr>
<td>SCK</td>
<td>“</td>
</tr>
<tr>
<td>Glassine</td>
<td>“</td>
</tr>
<tr>
<td>Poly-coated Kraft</td>
<td>“</td>
</tr>
</tbody>
</table>

Flame plasma surface treating systems either for Pre-Treatment or Post Treatment are designed to promote the adhesion of inks, coatings, laminations, etc.

Untreated polymers have a surface energy of 30 – 32 dynes/cm. In order to obtain acceptable adhesion, the difference in surface energy between the substrate and the topcoat should be at least 10 dynes/cm^1. 

There are several adhesion promoting technologies that are used to raise the surface energy of the substrate, namely:
Table 2: Comparison of Adhesion Promotion Technologies

<table>
<thead>
<tr>
<th>TECHNOLOGY</th>
<th>ADVANTAGES</th>
<th>DISADVANTAGES</th>
</tr>
</thead>
<tbody>
<tr>
<td>Priming/Coupling agents</td>
<td>Good results</td>
<td>Recurrent replacement cost, environmental concerns</td>
</tr>
<tr>
<td>Corona</td>
<td>Widely used</td>
<td>Limited line speed</td>
</tr>
<tr>
<td>DIRECT FLAME</td>
<td>Higher treatment levels, No limit to line speed</td>
<td>Higher initial cost</td>
</tr>
<tr>
<td>Atmospheric plasma</td>
<td>Good results</td>
<td>High cost for carrier, limited line speed</td>
</tr>
<tr>
<td>Ozone, UV, Ozone + UV</td>
<td>Used for 3-D substrates</td>
<td>Line speed limited to 30 m/min (98 fpm)</td>
</tr>
</tbody>
</table>
**Pre-Treatment**

The main use of a flame treater for pre treatment of paper board is to promote the adhesion of PE extruded onto the board, or to promote the adhesion of a primer onto the board prior to extrusion coating.

The primary function of the flame treater is to add heat to the board and to burn off the tiny paper fibers present on the board surface.

The critical parameter in pre treatment is the amount of heat applied to the board. Flame chemistry, which will be covered later is not critical, although more and more companies are using this parameter as a quality control measure.

The amount of heat applied is a function of burner sizing. For lines speeds of 1000 FPM (325 m/min) approximately 15,000 Btu/Hr/In (148 kcal/mm/Hr) is usually required.

Higher line speeds require more heat. The relationship, however, is not proportional, but involves such parameters as line speed, ambient conditions, paper weight, coat weight, etc.

The amount of heat generated by the burner is controlled by the combustion air pressure.
Post Treatment

There are five key variables that optimize flame plasma surface treating for post treatment are:

- Flame Chemistry
- Amount of Plasma Generated
- Flame Geometry
- Distance of the Substrate from the Flame
- Dwell Time

Flame Chemistry

Flame chemistry is determined by the air/fuel ratio. In general, the stoichiometric ratio is approximately 10 air:1 part gas, for natural gas. It is 24:1 for LPG and 32:1 for butane.

The optimum flame chemistry is that which provides for an O₂ concentration, in the flame plasma, that is, after the combustion reaction, of 0.1% – 0.5%\(^1\).

Simply setting the optimum air/fuel ratio will not assure optimal surface treatment for the following reasons. In the combustion industry, air and fuel are measured by volume, however, the combustion reaction is based on mass (gram or pound mole). Controlling air and gas on a volume basis does not take into account the change in O₂ concentration in the air due to combustion air temperature and humidity. A combustion air blower provides a fixed volume of air regardless of the air density, which will be affected by temperature and the humidity of the air. Higher temperature air contains less mass of O₂; if the air is humid, the H₂O will displace some of the oxygen. If the flame chemistry is optimized on a cool dry day, on a hot, humid day it may not at the optimal ratio.

Similarly, gas composition changes based on the gas field from which the gas is drawn. Again, optimizing at one gas composition does not mean optimal flame chemistry at a different composition.

The change in the O₂ concentration with air temperature and humidity is illustrated in Table 5. Table 6 illustrates the differences in gas composition based on the gas field.
Table 3: Effect of Temperature & Humidity on O₂ in Air

<table>
<thead>
<tr>
<th>Temperature (°F)</th>
<th>Humidity (%)</th>
<th>O₂</th>
<th>N₂</th>
<th>Ar₂, CO, H₂</th>
<th>H₂O</th>
<th>Density (lb./ft²)</th>
</tr>
</thead>
<tbody>
<tr>
<td>60</td>
<td>0</td>
<td>20.99</td>
<td>78.03</td>
<td>0.98</td>
<td>0</td>
<td>0.07632</td>
</tr>
<tr>
<td>60</td>
<td>80</td>
<td>20.70</td>
<td>79.64</td>
<td>0.97</td>
<td>1.4</td>
<td>0.07592</td>
</tr>
<tr>
<td>60</td>
<td>100</td>
<td>20.62</td>
<td>76.67</td>
<td>0.96</td>
<td>1.75</td>
<td>0.07581</td>
</tr>
<tr>
<td>90</td>
<td>20</td>
<td>20.79</td>
<td>0.97</td>
<td>0.97</td>
<td>0.95</td>
<td>0.07189</td>
</tr>
<tr>
<td>90</td>
<td>80</td>
<td>20.19</td>
<td>75.06</td>
<td>0.90</td>
<td>3.81</td>
<td>0.07111</td>
</tr>
<tr>
<td>90</td>
<td>100</td>
<td>19.99</td>
<td>74.32</td>
<td>0.90</td>
<td>4.76</td>
<td>0.07086</td>
</tr>
</tbody>
</table>

Table 4: Gas Composition from Different Gas Fields

<table>
<thead>
<tr>
<th>Component</th>
<th>Birmingham</th>
<th>Ohio</th>
<th>Pittsburg</th>
<th>Alaska</th>
</tr>
</thead>
<tbody>
<tr>
<td>Methane (CH₄)</td>
<td>90.0</td>
<td>94.1</td>
<td>83.4</td>
<td>99.6</td>
</tr>
<tr>
<td>Ethane (C₂H₆)</td>
<td>5.0</td>
<td>3.0</td>
<td>16</td>
<td>------</td>
</tr>
<tr>
<td>Propane (C₃H₈)</td>
<td>-----</td>
<td>0.4</td>
<td>-----</td>
<td>------</td>
</tr>
<tr>
<td>Volume air/Volume gas (Stoichiometric)</td>
<td>10.47</td>
<td>10.72</td>
<td>11.70</td>
<td>10.52</td>
</tr>
</tbody>
</table>

First generation flame treaters attempted to compensate for the changes in flame chemistry due to differences in air temperature, humidity and gas composition by the use of flow meters. However, flow meters cannot compensate for these differences in combustion air temperature and humidity since they measure volume, not mass flow. Changes in gas composition similarly cannot be compensated for by flow meters.

Because of these limitations, first generation flame treaters were not reliable, in terms of repeatability and consistency of treatment.

An in-line flame plasma analyzer was developed in the 1980’s as an effective control device to overcome these limitations. The analyzer operates as follows:

A sample of the air/gas mixture is fed into the first stage catalytic cell at a fixed rate. The catalytic cell is maintained at approximately 650 °C (1200 °F). The sample ignites, burns, and the products of combustion (POC’s) flow into the analyzer cell, maintained at a lower temperature and which incorporates a ceramic/platinum sensor produces an electrical voltage proportional to the
amount of excess $O_2$ or excess hydrocarbon present in the POC's. The signal is amplified, conditioned and linearized. A discrete step function occurs when there is no excess $O_2$ or excess hydrocarbon (unburnt fuel) is present in the POC's. This is the stoichiometric ratio of oxygen to fuel.

The signal is in used as an input to the PLC which has a control loop output to operate a gas flow control valve, open or close, to maintain the specified set point amount of excess $O_2$ in the POC's.

This value depends on the specific substrate, line speed and treatment level required and is best determined by trials

A schematic representation of the analyzer is shown in figure 1.

**Figure 1: Flame Plasma Analyzer**
In the most recent development, control of the flame chemistry has been enhanced by the use of mass flow meters for both air and gas together with the use of a flame analyzer, used to measure and control the amount of excess hydrocarbon (gas) or O₂ in the flame plasma. This arrangement allows for very fast response time as well as for real time calibration of the mass flow of combustion air and fuel gas. This is critical when the line starts up at 100m/min (328 fpm) and quickly ramps up to 1000m/min (3280 fpm). The ability to quickly achieve optimum flame chemistry minimizes production losses.

**Plasma Generation**

Critical issues in the design of the flame treater for post treatment are:

I) Burner design to prevent laning or striations.

II) Burner design and control scheme for operation at line speeds of 100m – 1000m/min

III) Combustion noise

Laning is defined as a periodic variation in the treatment level of a polymer substrate in the cross machine direction. Researchers have indicated that laning is due to the spacing of the flames in the treatment burner and the firing rate of the burner.

There are two types of burners used for flame treating.

Ribbon burners consist of a series of stainless steel “ribbons” the crimp and angle of which are specifically designed for the type of substrate & substrate geometry inserted in a high capacity extruded aluminum burner, to insure uniformity of treatment and totality of treatment.

A ribbon stack is shown in Figure 2. Figures 3 and 4 show a cross section of a ribbon burner and a photo of the (8) port wide slot burner that was used in this high speed installation, Fig. 4.
Figure 2: Typical Ribbon Burner Stack
The second burner type is the drilled port burner which consists of a CNC drilled port brass inserted in an aluminum body is shown in figure 5.
Ribbon burners are inherently backfire proof. This means that if the velocity of the air gas mixture is less than the rate of flame propagation, the flame will not travel back to the burner manifold and burn. This is achieved in a ribbon burner by the cooling effect on the flame as it travels backward through the ribbon stack. The flame is cooled below minimum flame temperature to sustain ignition.

The manufacturer of the drill port burner makes a similar claim.

To address these three issues, burner design to prevent laning, design and control philosophy for operation over a wide range of machine speeds and combustion noise, a ribbon burner was designed in a wide slot configuration which proves sufficient plasma at 1000 m/min and the flame geometry to eliminate striations.

A comparison of a 2 port narrow slot ribbon burner with the 8 port wide slot is shown in figure 6.
Figure 6: Schlering Image of (2) Port Ribbon Burner compared to (8) Port Wide Slot

(2) Port Ribbon Pattern  (8) Port Ribbon pattern

It is apparent from these Schlering images that at high line speeds the (2) port ribbon does not provide as much plasma as does the (8) port. More importantly, the (8) port geometry is such that laning is minimized when compared to a (2) port firing at the same rate.

At high line speeds, those in excess of 800 m/min (2624 fpm), an alternative to the wide slot burner is the use of two or more burners. This adds cost, unnecessary complexity and the need for additional maintenance to the flame treater.

Another critical design issue that had to be dealt with for this project was how to operate the treater from a line speed of 100 m/min (328 fpm) up to 1000m/min (3280 fpm). At low line speeds, the plasma output required is approximately 100 Kcal/Hr/cm (1000 Btu/Hr/In), at high line speeds as much as 3000 Kcal/Hr/cm (30,000 Btu/Hr/In).

One approach is to use multiple burners (at least 4, perhaps as many as 6), this would be the case if a drilled port burner would have been used, since the drilled port burner is not capable of providing sufficient output without laning and might not even be able to treat the substrate, at the specified high end line speeds.
However, the use of an (8) port wide slot burner with a unique air/gas mixing control strategy allowed the use of only two burners, firing at 50% of maximum capacity at 1000 m/min. This provides the capability to increase line speed 30% in the future without any change to the flame treating system.

When firing a burner, whether a ribbon burner or a drilled port burner, against a back up roll at a distance of 6-12.5 mm (1/4” – 1/2”), combustion harmonics or noise, is usually manifested. The resolution of noise is difficult and this is where experience in flame treating is critical, knowing what to do to eliminate the noise.

The proper design of the exhaust system is critical in eliminating and/or minimizing combustion noise. Treater design to interfere with the resonance created by the combustion of the air/gas mixture plays a critical role in noise suppression.

The elimination or at least the attenuation of the noise is usually accomplished by a variation in the ribbon pattern and/or the interruption of the resonating column by insertion of baffles or other means.

**Flame Geometry**

In order to obtain uniformity of surface treatment, the plasma must react with the surface of the substrate evenly, without discontinuity. This uniformity of treatment is achieved by use of a ribbon burner, which provides an extremely uniform flame front geometry, without laning.

**Distance of the Substrate from the Flame**

Average distance of the substrate to the flame varies from 6 – 12.5 mm (1/4” – 1/2”). The optimum distance is a result of field trials and in this project turned out to be 8mm. However, noise and laning must be considered in determining the optimum distance.

**Dwell Time**

The substrate needs to be in contact with the flame plasma for enough time to affect the surface. This is achieved by providing a burner that can produce
sufficient plasma in a uniform manner. At line speeds of 1000 m/min (3280 fpm), the burner must be capable of providing 3000 Kcal/Hr/cm (30,000 Btu/Hr/In) without laning.

A single (8) port ribbon burner can provide this level of plasma generation, uniformly and without laning.

A schematic of the web path through the flame treater is shown in figure 7. The burner, nip roll, water cooled back up roll, heat shield and idler rolls are shown.

A photo of the flame treater is shown in figure 8.
Figure 7: Web path Through Flame Treater Station
Figure 8: High Speed Flame Treater
Bibliography


What's it all about---Adhesion Promotion using Flame Plasma

Presented by:
Joe DiGiacomo
Sales Director
Flynn Burner Corporation
What’s Flame plasma treatment?
It’s NOT......
Flame Plasma Treatment is...

Using the **reactive species** in the products of combustion (flame plasma) to **increase** the **surface energy** of the substrate
Reactive Species in the Flame Plasma

- $(\text{OH}^-)$ - Hydroxyl Free Radical
- $(\text{COOH}^-)$ - Carbonyl
- $(\text{O}=\text{C}-\text{O})$ - Carboxyl
- $(\text{R}-\text{O}-\text{R})$ - Ether
- $(\text{R}-\text{C}=\text{O}-\text{OR}')$ - Ester
Reactive Species in the Flame Plasma

(O) + (O⁻) - Oxygen atoms & ions

(e⁻) - Free electrons
What is adhesion promotion all about ?????
“Kevin and his Dad are bonding.”
## Comparison of Adhesion Promotion Technologies

<table>
<thead>
<tr>
<th>TECHNOLOGY</th>
<th>ADVANTAGES</th>
<th>DISADVANTAGES</th>
</tr>
</thead>
<tbody>
<tr>
<td>Primers</td>
<td>Excellent results</td>
<td>Replacement cost, Environmental issues</td>
</tr>
<tr>
<td>Corona</td>
<td>Widely used</td>
<td>Limited line speed</td>
</tr>
<tr>
<td>DIRECT FLAME</td>
<td><strong>No limit to line speed</strong></td>
<td><strong>Higher initial cost</strong></td>
</tr>
<tr>
<td>Atmospheric plasma</td>
<td>Good for hard to treat substrates</td>
<td>Cost &amp; availability of carrier gas</td>
</tr>
</tbody>
</table>
Flame, Corona & Atmospheric Plasma

- **POLARIZE** the surface of the substrate by:

- **OXIDIZING** the surface by adding “functional groups”

- **ALTERING** the electron distribution & density AT THE SURFACE
Reactive Species in the Flame Plasma

- $(\text{OH}^-)$ - Hydroxyl Free Radical
- $(\text{COOH}^-)$ - Carbonyl
- $(\text{O=C-O})$ - Carboxyl
- $(\text{R-O-R})$ - Ether
- $(\text{R-C=O-OR}')$ - Ester
Reactive species in the POC’s (Plasma)

- (OH\textsuperscript{-}) - Hydroxyl free radical
- (COOH\textsuperscript{+}) - Carboxylic
Pre-Treatment---What is it?

Application of flame to a substrate, usually paper board, prior to extrusion coating
Key Variables in Pre-Treatment

- Flame geometry
- Amount of plasma (heat) generated
- Distance of the burner from the part
- Flame Chemistry
Key Variables in Pre-Treatment

- Flame geometry

  Provides uniformity of treatment across the sheet
Key Variables in Pre-Treatment

Amount of plasma (heat) generated

1) Burns off the small fibers &
2) Hot board enhances PE adhesion
Key Variables in Pre-Treatment

- **Distance of the burner from the substrate**
  - ~1/2” (12mm) – 3” (75mm)
Control System For a Pre-treater
Zero Gas Pressure Regulator w/
Manual Air Control
Key Variables in Pre-Treatment

- Flame Chemistry

Adds reactive species to the surface of the board which enhances PE adhesion
Control System
ZERO REGULATOR WITH PLASMA ANALYSIS
Post Treatment---What is it?

Application of flame to:

- PE surface of paper board
- Directly to polymer films, (BOPP, OPP, foils, laminated structures, complex film structures)
Key Variables in Post Treatment

- Flame chemistry (air/gas ratio ~ 10:1)
- Flame geometry
- Amount of plasma generated
- Distance of the burner from the part
Flame Chemistry

Optimum flame chemistry

0.1 – 0.5% $\text{O}_2$ in the flame plasma (in the post combustion gases)
Flame Chemistry

**Figure 1**

**Effect of Air/Gas Ratio on Treatment Level**

FUEL - METHANE
CONVEYOR SPEED - 75 ft/min (22.8 m/min)
DISTANCE FROM FLAME - 1/2 IN. (12.7 mm)
Flame Chemistry

**Figure 1A**

EFFECT OF AIR/GAS RATIO ON TREATMENT LEVEL

FUEL-PROPANE
CONVEYOR SPEED - 75 ft/min (22.8 m/min)
DISTANCE FROM FLAME - 1/2 IN. (12.7 mm)
What affects flame chemistry ??

- Combustion
- Air

- Fuel
Effect of temperature & humidity on O$_2$ in combustion air

<table>
<thead>
<tr>
<th>TEMPERATURE</th>
<th>Humidity (%)</th>
<th>O$_2$ (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>°C (°F)</td>
<td></td>
<td></td>
</tr>
<tr>
<td>16 (60 °F)</td>
<td>0</td>
<td>21</td>
</tr>
<tr>
<td>32 (90 °F)</td>
<td>100</td>
<td>20</td>
</tr>
</tbody>
</table>
## Gas composition from different gas fields

<table>
<thead>
<tr>
<th>COMPONENT</th>
<th>North Sea</th>
<th>Algeria</th>
<th>Alaska</th>
</tr>
</thead>
<tbody>
<tr>
<td>Methane $\text{CH}_4$</td>
<td>90%</td>
<td>94%</td>
<td>99%</td>
</tr>
<tr>
<td>Ethane $\text{C}_3\text{H}_6$</td>
<td>5%</td>
<td>3%</td>
<td>------</td>
</tr>
<tr>
<td>Stoichiometric ratio</td>
<td>10.5</td>
<td>10.7</td>
<td>10.5</td>
</tr>
</tbody>
</table>
Flame Plasma Analyzer

Diagram of Flame Plasma Analyzer:
- PLC
- Cell control and signal processing electronics
- Analyzer Cell
  - Heater
  - T/C
  - Sensor
- Sample of air/gas mixture
- Flame Plasma Burner
- Motorized Control Valve
- Zero Pressure Gas
- Combustion Air
- Deviation Alarm
Flame Plasma Analyzer
A other Method to control flame chemistry

Flame Temperature

Measures the temperature of a small pilot flame and uses an algorithm relating flame temperature to air/fuel ratio
Flame Geometry & Amount of Plasma Generated

Determined by:

- Burner design
- Ribbon
- Drilled port
Flame Geometry

And....it’s critical because it affects

THE UNIFORMITY of
TREATMENT
(Laning/Striations)
Flame Geometry Studies

0 speed

2 PORT

8 PORT
Flame Geometry Studies

Low

2 PORT

8 PORT
Flame Geometry Studies

High

2 PORT

8 PORT
Burner Types

Ribbon

Drilled Port
Aluminum Burner
(8) Port
Aluminum Burner
(4) Port
Drilled Port
Advantages of Ribbon vs. Drilled Port Burner

RIBBON BURNER PROVIDES:

- More Energy
  (up to 200 kcal/mm/20,000BTU/In) without striations

- More Uniform Treatment
Distance Burner to Part
Distance - Burner to Part

Figure 2

Effect of Distance from Part to Inner Cones of Flame

Distance (inches)

10.37 Air/Gas Ratio
.02 Sec. Contact Time
Fuel - Methane
Control System
Zero Gas Pressure Regulator w/ Manual Air Control

[Diagram of a control system with labels such as 'Fuel Gas Supply @1PSI', 'Air Flow Manual Butterfly Valve', 'Balanced Zero Regulator', 'Impulse Line', 'Venturi Mixer with Ratio Setting Screw 1-2" W.C.', 'Air/Gas Mixture to Burner']
Control System
Mass Flow Control
Control System
Mass FLOW CONTROL with PLASMA TRIM
Flame Treatment—it’s about.....

PROMOTE ADHESION
Considerations for a Flame Treater

- Control of flame chemistry
- Burner selection
- Distance
- Control philosophy
Thank You

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Please remember to turn in your evaluation sheet...