



# Aqueous poly(lactic acid) copolymer dispersions as paperboard barrier coatings

Presented by:

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## Aqueous (bio)polymer dispersions

- Polymer particles in nano- to micrometer size dispersed in water. During drying these particles coalesce into a uniform polymer layer.
- Highlights:
  - Enables thinner coatings than typically possible by extrusion,
  - Alternative application techniques (on- and off-machine, printing presses),
  - Freedom for preparing formulations with different additives.
- Lowlights:
  - Majority of current products are at least partly petroleum based,
  - Only few products provide all the required properties alone,
  - Recycling performance and viable end-of-life options to be clarified,
  - Positioning in EU directives/regulations for specific applications.



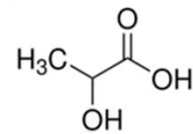
## Cross-linkable poly(lactic acid) dispersions

- Typically PLA dispersions not feasible due to hydrolytic instability,
- Goal is to overcome this challenge:
  - Unsaturated cross-linkable copolymers prepared by bulk polycondensation,
  - Bio-based content of the polymer up to 100%,
  - Biodegradation in water is faster than with neat PLA,
  - Dispersions with the thermomechanical method with PVOH as a stabilizer,
  - Pilot coated dispersion has outperformed extruded PLA coatings.
- Applications include also self-healing materials for surface treatment and encapsulation of hydrophobic active compounds

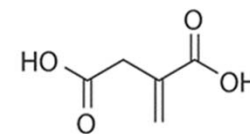
<https://aaltodoc.aalto.fi/handle/123456789/23472>

Mehtiö, T., J. App. Pol. Sci., vol. 134, issue 1, p. 44326 (1-8).

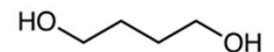
Rämö, V. Eur. Pol. J., vol. 48, issue 8, p. 1495–1503.



D,L - lactic acid



Itaconic acid



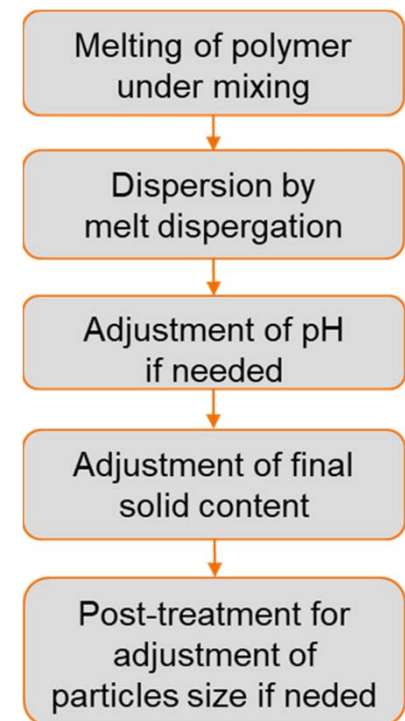
1,4 - butanediol



## Thermomechanical method

- Available, scalable, and environmentally feasible, and referred to also as solvent-free extrusion emulsification,
- Exposing a blend of thermoplastic polymer, stabilizer and water to intensive shear forces at elevated temperatures,
- Equipment, process parameters and additives chosen based on the thermal behavior, melt viscosity, and degradation tendency of the polymer. High  $T_m$ , for example, requires pressurized equipment,
- Requires that molten polymer has low viscosity for transferring polymer to the aqueous phase under shear during dispersion preparation,
- Used for both bio- and petroleum-based polymers.

### Example:





## Copolymers and dispersions (not cross-linked)

### TYPE 1:

- New monomers for functionality,
- Decreased amount of partially (low) saponified PVOH as a stabilizer.
- Dispersions produced using glass and/or 2L Juchheim reactor.

| Copolymer | Mn (g/mol)         | Mw (g/mol) | Tg (°C)   |
|-----------|--------------------|------------|-----------|
| 1.1       | 276                | 1 890      | 24 ± 1    |
| 1.2       | 1 580              | 8 280      | 5.7 ± 0.2 |
| 2.1       | 3 130              | 21 170     | 41 ± 0.7  |
| 2.2       | N.A. in chloroform |            | 28 ± 0.4  |
| 2.3       | 6 200              | 7 840      | 39 ± 0.2  |

### TYPE 2:

- New type of monomers,
- Partially saponified PVOH as a stabilizer.
- Polymers / dispersions produced in pilot scale with a Lödige DVT 10L reactor and post-treated.



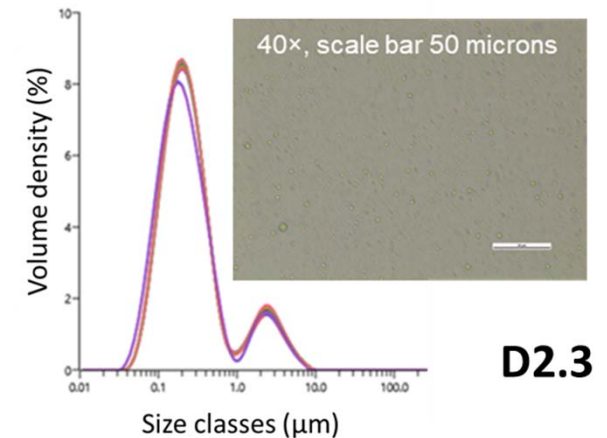
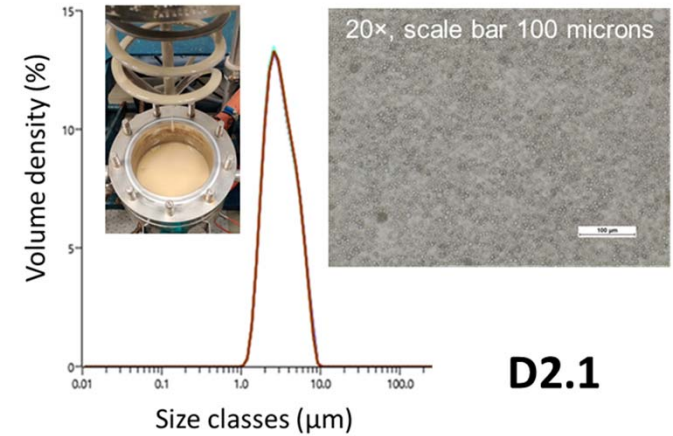
<https://www.inn-pressme.eu/>



## Characteristics of dispersions

| CODE | POLYMER | VISCOSITY (100 RPM) | SOLIDS CONTENT | PARTICLE SIZE         |
|------|---------|---------------------|----------------|-----------------------|
| D1.1 | 1.1     | 150 mPas            | 29%            | 2-40 $\mu\text{m}$    |
| D1.2 | 1.2     | 260 mPas            | 34%            | 1.5-15 $\mu\text{m}$  |
| D2.1 | 2.1     | 830 mPas            | 27%            | 1-10 $\mu\text{m}$    |
| D2.2 | 2.2     | 1100 mPas           | 40%            | 1-10 $\mu\text{m}$    |
| D2.3 | 2.3     | 630 mPas            | 35%            | 0.04-10 $\mu\text{m}$ |

pH of dispersions in the range of 1-3





## Reference polymer dispersions

|              | CODE | TYPE  | CHARACTERISTICS                      |
|--------------|------|---|--------------------------------------|
| Conventional | A    | Formulated with wax, solids 46%                           | Hydrophobicity, WVTR                 |
|              | B    | Styrene butadiene, $T_g < 15\text{ }^\circ\text{C}$       | Water vapor barrier                  |
|              | C    | All Acrylic, $T_g < 15\text{ }^\circ\text{C}$             | Oil & grease resistance              |
| Novel type   | D    | Poly(lactic acid) based, MFFT $20\text{ }^\circ\text{C}$  | Particle size $1\text{ }\mu\text{m}$ |
|              | E *  | Poly(lactic acid) based, MFFT $160\text{ }^\circ\text{C}$ | Particle size $5\text{ }\mu\text{m}$ |
|              | F    | 70 : 30 blend of D : E                                    |                                      |
|              | G    | Polyethylene based, solids 50%                            | Water vapor barrier                  |
|              | H    | Polyester based, solids 40%                               | Water barrier                        |

\* Tested only in a blend



## Laboratory coating

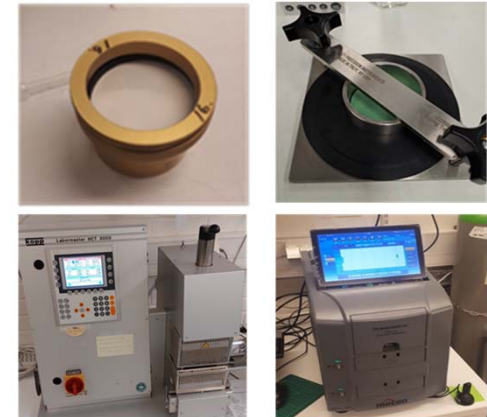
- Two paperboard substrates;
  - Smooth surface of 170 g/m<sup>2</sup> cupstock for prescreening references
  - Coated reverse side of 210 g/m<sup>2</sup> folding boxboard for actual comparison.
- Coating carried out with ERICHSEN lab sheet coater at 50 mm/s,
- Drying performed in an air circulation lab oven at 100 °C for 5 min,
- Two target coat weights (15 & 20 g/m<sup>2</sup> and 11-12 & 15-16 g/m<sup>2</sup>)
  - Coat weight adjustment with the metering rods and/or by coating solids
- Brookfield viscosity of all coating formulations measured,
- At least 10 sheets coated for each coating and coat weight combination.





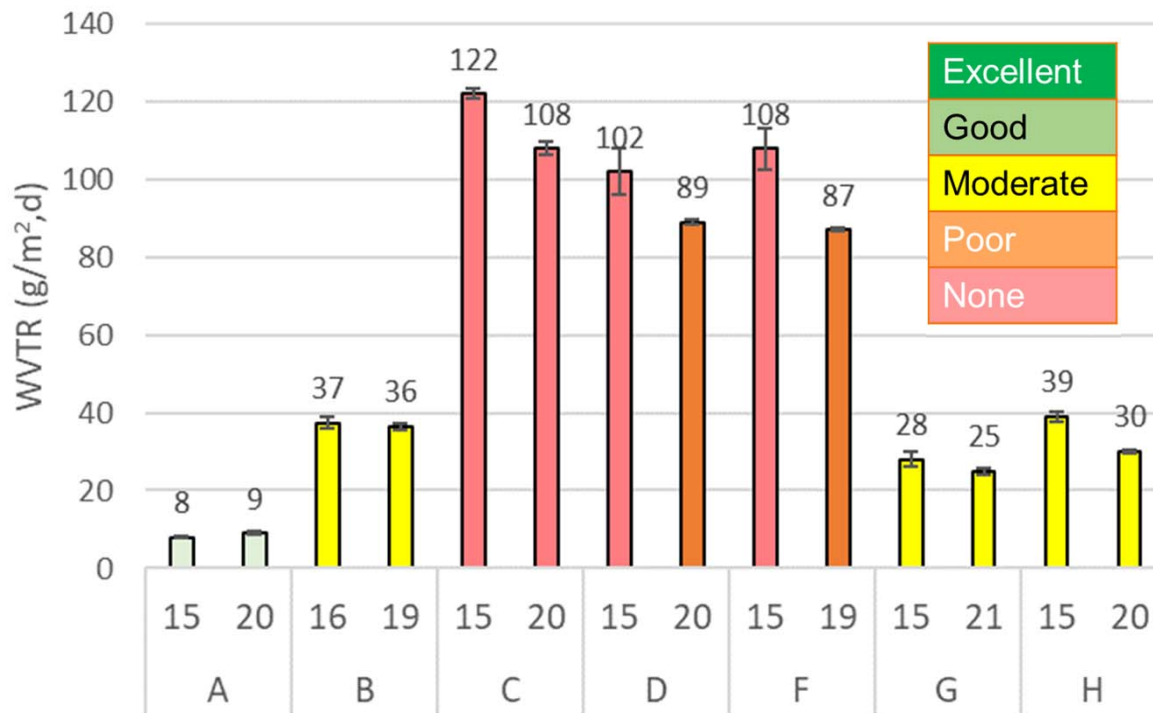
## Characterization of coated sheets

- Coat weight ( $\text{g}/\text{m}^2$ ) – difference in dried uncoated and coated board,
- Pinholes tested with dyed aqueous or alcohol-based liquids,
- Water absorption ( $\text{g}/\text{m}^2$ ) during 10 min (ISO 535:2014, TAPPI T-441),
- WVTR ( $\text{g}/\text{m}^2/\text{d}$ ) at 23 °C / 50% RH and 38 °C / 90% RH (ASTM F1249–13),
- OTR ( $\text{cm}^3/\text{m}^2/\text{d}$ ) at 23 °C / 0% RH (ASTM D3985),
- KIT test for OGR (TAPPI T-559),
- Grease and oil barrier (VTT Method),
- Hexane vapor transmission rate HVTR ( $\text{g}/\text{m}^2/\text{d}$ ),
- Heat sealability with visual ranking,
- SEM imaging.





## Water vapor barrier at 50% RH and 23 °C

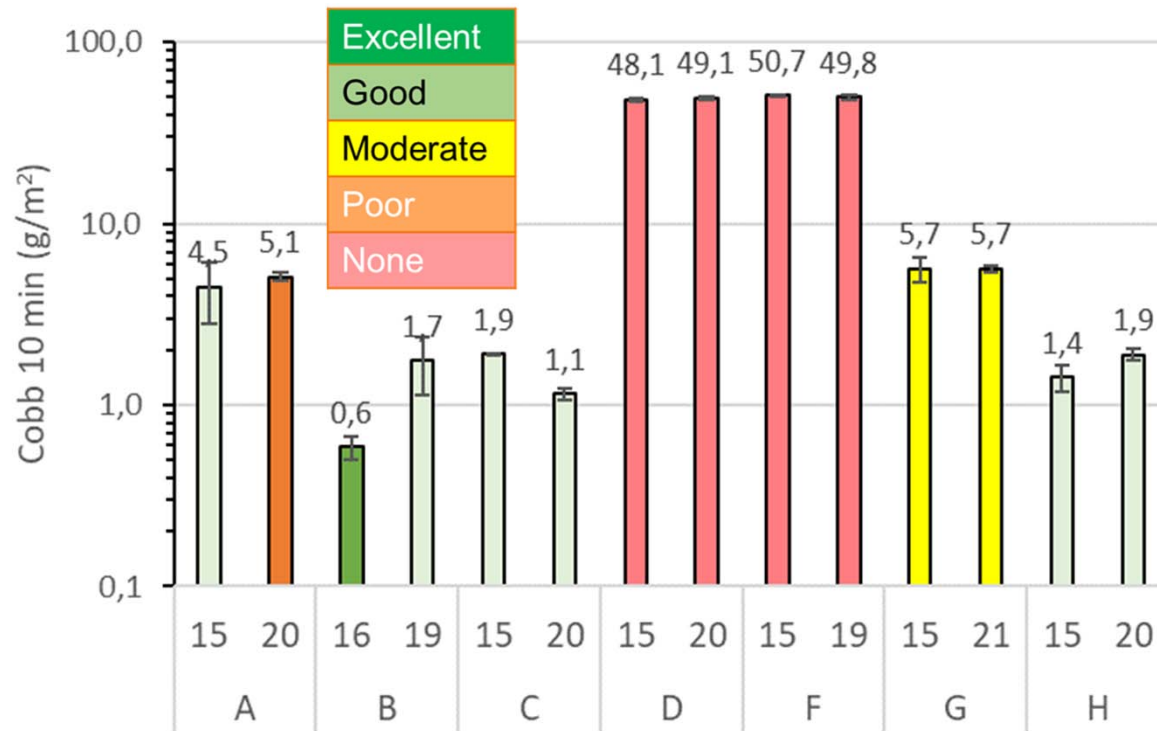


**Base board could not be measured**

- Dispersions intended for WV barrier had good to moderate barrier, and the best WVTR with wax formulated dispersion A,
- No clear correlation between water and water vapor barrier,
- PLA-based reference coatings performed slightly better than C,
- Increasing coat weight typically decreased WVTR, although the impact was not significant.



## Water absorption after 10 minutes



**Base board  $46 \pm 0,3 \text{ g/m}^2$**

- Some references good or even excellent water barriers.
- PLA references D and F did not form barrier at these drying conditions and coat weights,
- Above certain coat weight no obvious correlation with barrier performance,
- A limited number of defects (B) do not necessarily destroy water barrier.



## Grease and oil barrier

| Coating | g/m <sup>2</sup> | Breakthrough time * |
|---------|------------------|---------------------|
| A       | 15 & 20          | 6-24 HOURS          |
| B       | 16               | < 1 HOUR            |
|         | 19               | 6-24 HOURS          |
| C       | 15 & 20          | > 1 WEEK            |
| D       | 15               | 3-4 DAYS            |
|         | 20               | > 1 WEEK            |
| F       | 15 & 19          | > 1 WEEK            |
| G       | 15 & 21          | 1.25-2 DAYS         |
| H       | 15 & 20          | 6-24 HOURS          |

\* Breakthrough when at least two out of five pieces show  $\geq 1\%$  stained area

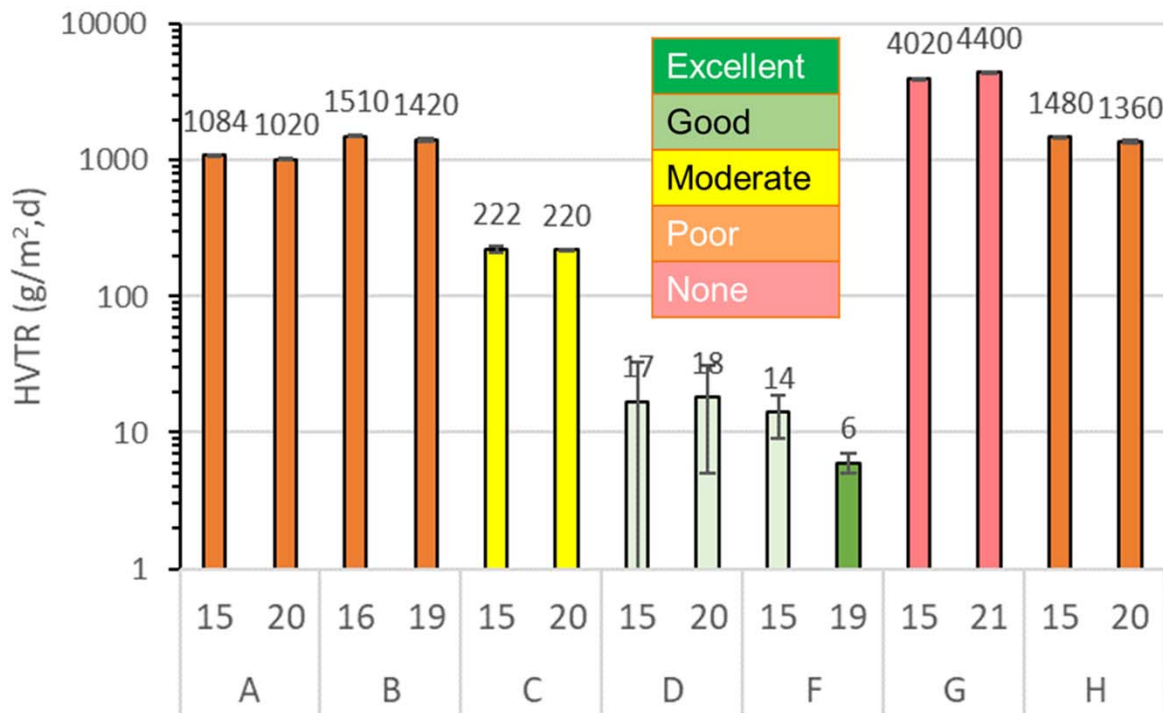
## Base board immediate staining

- Dispersion intended for grease barrier (C) and PLA based coatings (D,F) excellent barriers,
- Dispersions intended for water and/or water vapor barrier were at best moderate grease barriers,
- Grease barrier is sensitive to coating defects, which explains partly the poor performance of e.g. B

|           |
|-----------|
| Excellent |
| Good      |
| Moderate  |
| Poor      |
| None      |



## n-Hexane vapor (mineral oil) barrier



Base board  $7840 \pm 70 \text{ g}/(\text{m}^2, \text{d})$

- PLA based dispersions D and F provided the best barrier against n-hexane vapor,
- C with good grease barrier had better performance than the other two conventional dispersions A and B,
- PE based G as a polyolefin was the poorest barrier against n-Hexane



## Prescreening of reference dispersions

++ Good / Excellent

+ Moderate

0 Poor / None

|          | WATER | WATER VAPOR | N-HEXANE | GREASE | OXYGEN | SEALING      |
|----------|-------|-------------|----------|--------|--------|--------------|
| <b>A</b> | ++/+  | ++          | 0        | +      | 0      | <b>YES</b>   |
| <b>B</b> | ++    | +           | 0        | +/0    | 0      | <b>YES *</b> |
| <b>C</b> | ++    | 0           | +        | ++     | 0      | <b>YES</b>   |
| <b>D</b> | 0     | 0           | ++       | ++     | 0      | <b>YES</b>   |
| <b>F</b> | 0     | 0           | ++       | ++     | +      | <b>YES</b>   |
| <b>G</b> | +     | +           | 0        | ++     | 0      | <b>YES</b>   |
| <b>H</b> | ++    | +           | 0        | +      | 0      | <b>YES *</b> |

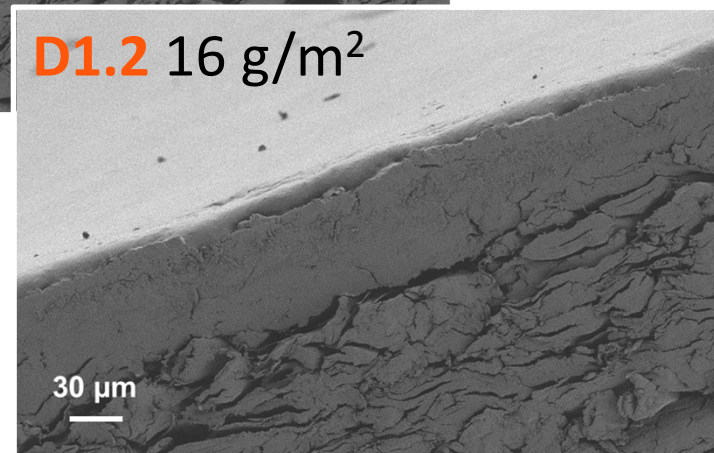
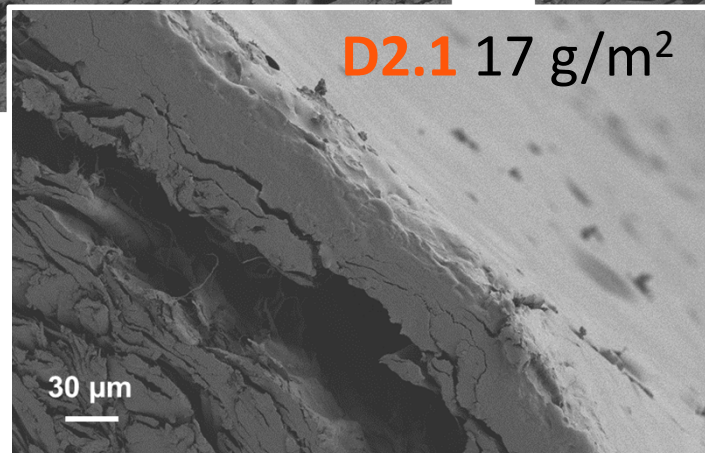
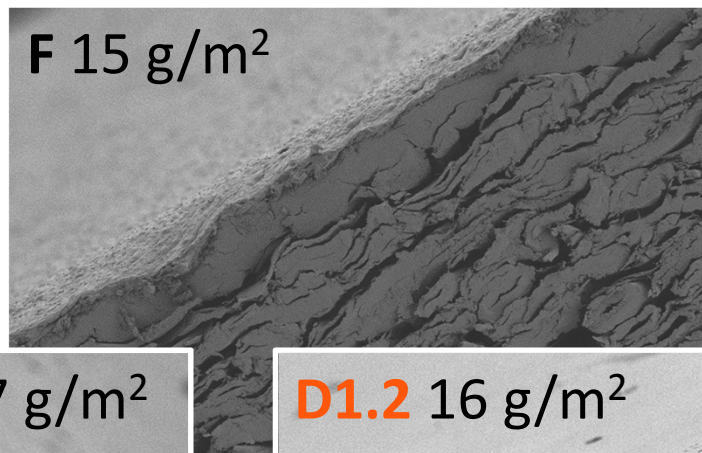
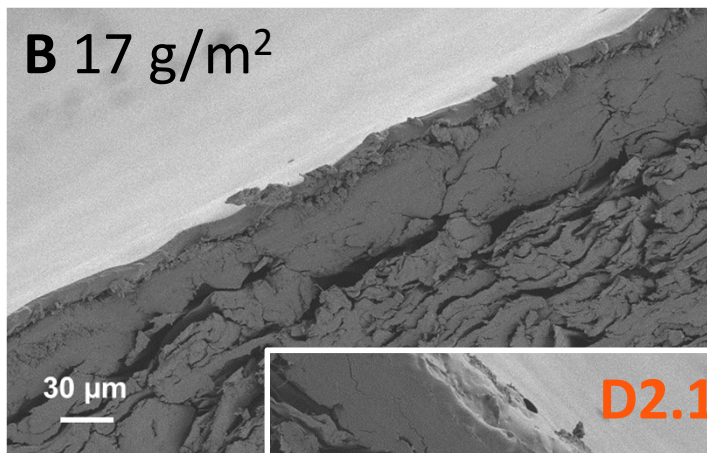


**Chosen for comparison with experimental dispersions**

\* Blocking tendency

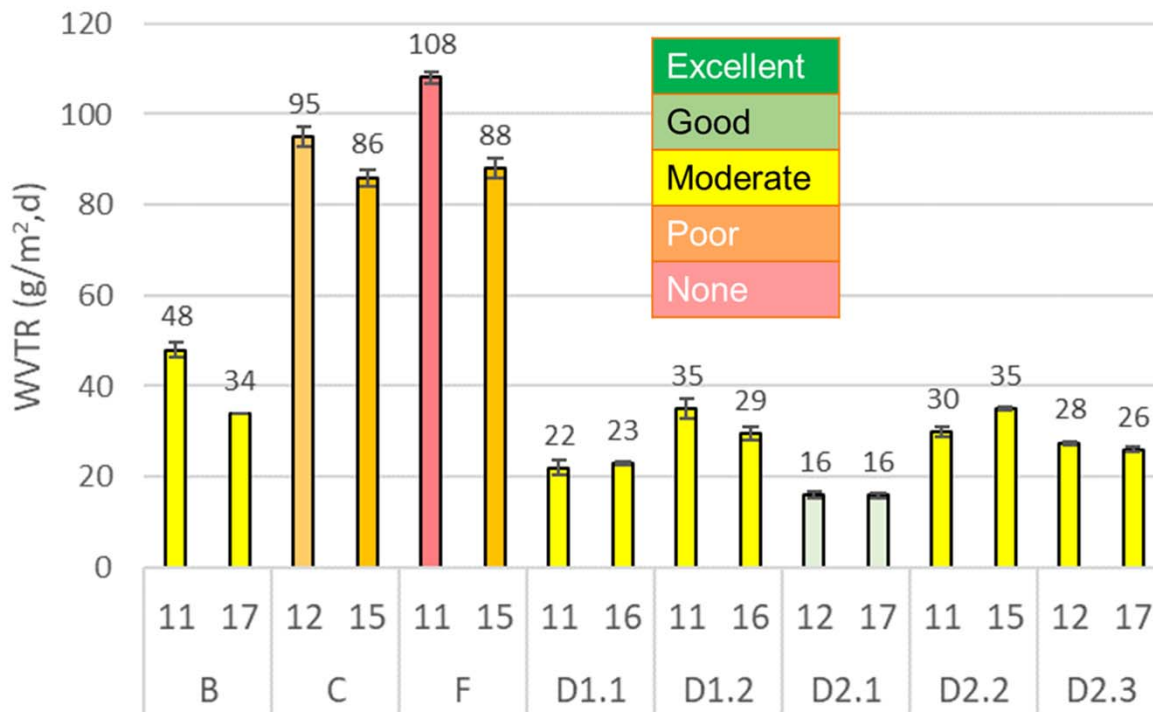


## Uniform poly(lactic acid) copolymer coatings





## Water vapor barrier at 50% RH and 23 °C

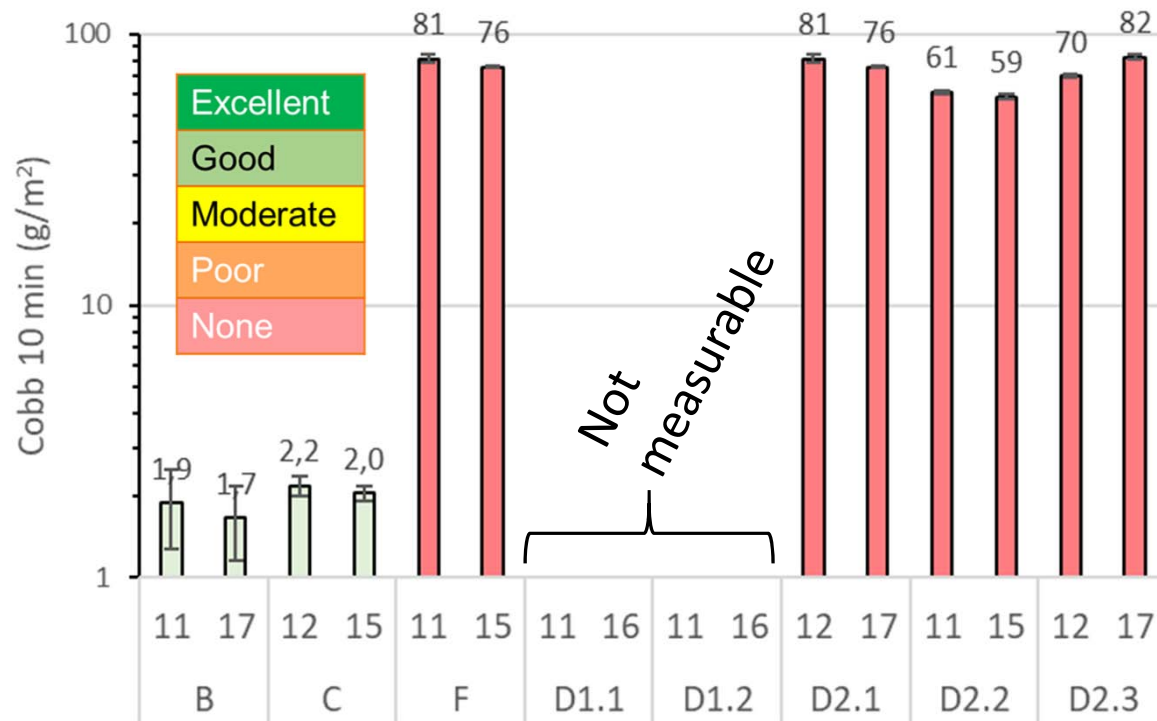


Base board 250 ± 10 g/(m<sup>2</sup>,d)

- Better than the reference PLA and similar/better than dispersion B,
- D2.1 the best (TYPE 2), and D1.1 better than D1.2 (TYPE 1),
- At tropical conditions TYPE 1 dispersions performed better than TYPE 2, but the values in general were significantly higher than with dispersion B likely due to PVOH used as a stabilizer.



## Water absorption after 10 minutes



**Base board  $110 \pm 1 \text{ g/m}^2$**

- B and C provided similar values also at lower coat weight with this substrate,
- Poly(lactic acid) dispersions - similarly to the reference F – had limited impact on barrier,
- TYPE 1 dispersions coatings could not be measured as the board delaminated, which is likely due to a more soluble PVOH and a low molecular weight polymer.



## Grease and oil barrier

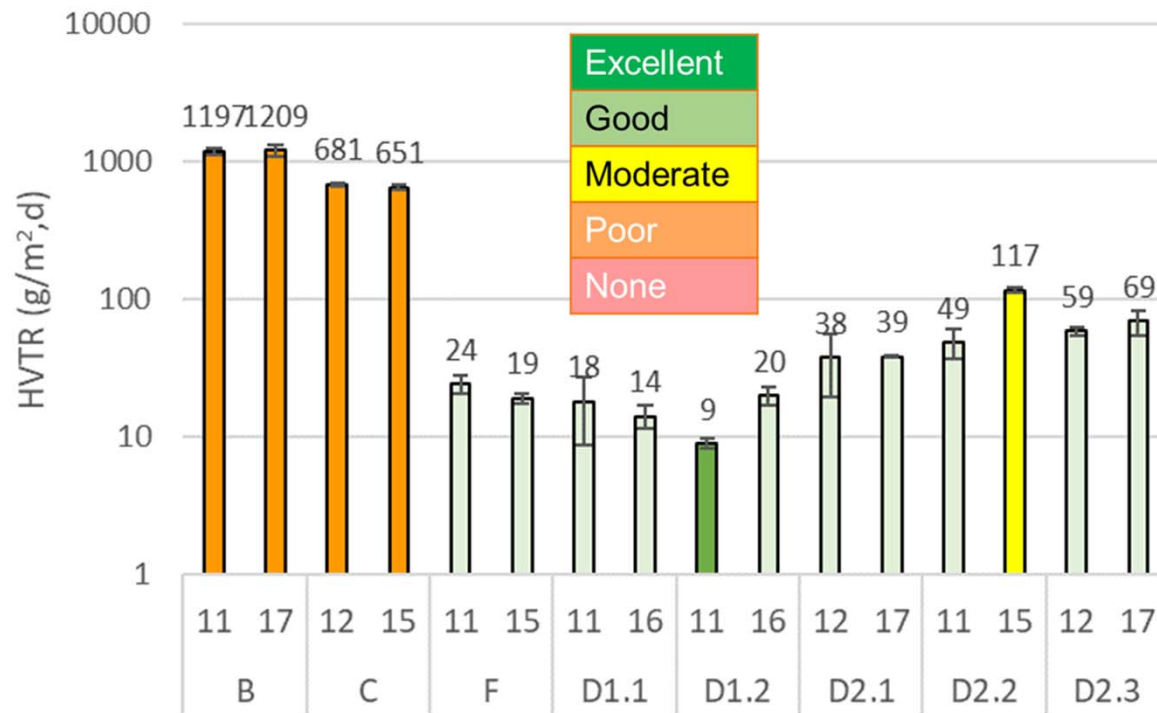
| Coating | g/m <sup>2</sup> | Breakthrough time * |
|---------|------------------|---------------------|
| B       | 11               | 3-4 DAYS            |
|         | 17               | 4-7 DAYS            |
| C       | 12 & 15          | > 1 WEEK            |
| F       | 11               | 6-24 HOURS          |
|         | 15               | > 1 WEEK            |
| D1.1    | 11 & 16          | > 1 WEEK            |
| D1.2    | 11 & 16          | 2-3 DAYS            |
| D2.1    | 12 & 17          | > 1 WEEK            |
| D2.2    | 11               | > 1 WEEK            |
|         | 15               | 4-5 DAYS            |
| D2.3    | 12               | 3-4 HOURS           |
|         | 17               | 5-6 HOURS           |

|           |
|-----------|
| Excellent |
| Good      |
| Moderate  |
| Poor      |
| None      |

- Dispersion C and most of the poly(lactic acid) dispersions were good barriers without staining even after one week,
- D2.1 had also maximum KIT value (12) at both coat weights,
- In spite of some scatter between the individual test pieces, D2.3 was poor barrier, and D1.2 was somewhat poorer than D1.1,
- Several new copolymer dispersions, however, provided better barrier than the reference F at low coat weight.



## n-Hexane vapor (mineral oil) barrier



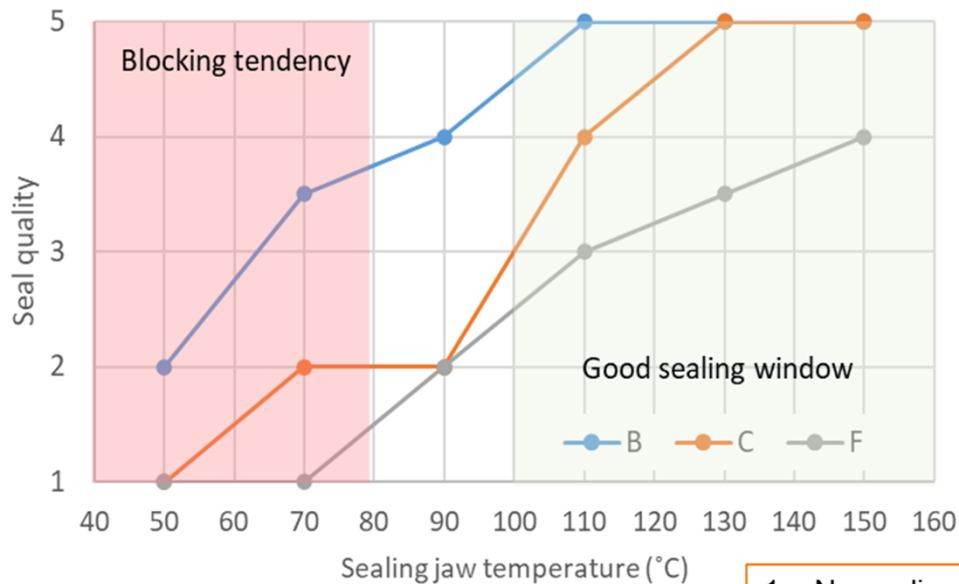
Base board 1200 ± 70 g/(m<sup>2</sup>,d)

- Both the PLA reference F and all the new dispersions performed better than the conventional references,
- TYPE 1 versions were on average similar to the reference F and better than TYPE 2 dispersions,
- Among the TYPE 2 dispersions D2.1 was slightly better than D2.2 or D2.3.

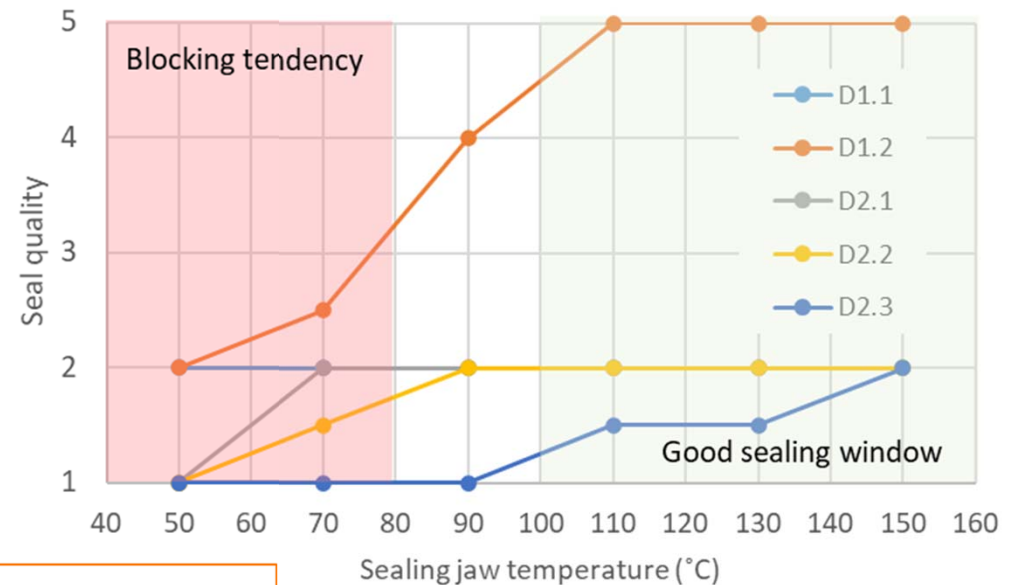


## Heat sealing (higher coat weight)

Reference dispersions



Novel lactic acid-based dispersions



- 1 – No sealing
- 2 – Weak seal (with sound)
- 3 – < 50% seal area damaged
- 4 – ≥ 50% seal area damaged
- 5 – complete damage (in board)

**Settings: 200 N and 1.5 s**  
**Coating against coating.**



## Performance of different dispersions

++ Good / Excellent

+ Moderate

0 Poor / None

|             | WATER | WATER VAPOR | N-HEXANE | GREASE | OXYGEN | SEALING |
|-------------|-------|-------------|----------|--------|--------|---------|
| <b>B</b>    | ++    | +           | 0        | ++     | 0      | + *     |
| <b>C</b>    | ++    | 0           | 0        | ++     | 0      | ++      |
| <b>F</b>    | 0     | 0           | ++       | ++/+   | 0      | ++      |
| <b>D1.1</b> | 0     | +           | ++       | ++     | 0      | 0       |
| <b>D1.2</b> | 0     | +           | ++       | ++     | 0      | + *     |
| <b>D2.1</b> | 0     | ++          | ++       | ++     | 0      | 0       |
| <b>D2.2</b> | 0     | +           | ++/+     | ++     | 0      | 0       |
| <b>D2.3</b> | 0     | +           | ++       | 0      | 0      | 0       |

\* Blocking tendency



## Conclusions

- Poly(lactic acid) copolymer dispersions with solids content up to 40% produced;
  - Production of polymers and dispersions upscaled from lab to pilot scale,
  - Post treatment (filtration, mechanical treatment) can have positive impact on particle size,
- Performance to be optimized with copolymer composition and additives, and a precoated substrate recommended for optimal coating quality,
- Tested non-crosslinked dispersions feasible for dry and greasy products;
  - Excellent oil and grease resistance and a good barrier against mineral oil migration,
  - Water vapor barrier at 50% RH comparable to a dispersion designed for WP barrier,
  - Applications with a direct water contact or high humidity challenging,
  - None of the dispersions (including references) provided true oxygen barrier,
- Heat sealing performance to be improved;
  - Dispersion from a low  $T_g$  copolymer indicated the possibility for adjusting heat sealability,
- Findings to be verified in pilot scale.



## Packaging concepts and demonstrations



|                    |
|--------------------|
| PLA DISPERSION     |
| FIBRIL. CEL. COAT. |
| PLA DISPERSION     |
| BASE PAPER         |

} ~ 38 g/m<sup>2</sup>

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Kumar, Vinay, 18th Biennial TAPPI European PLACE Conference | 10-12 October 2022



|                |
|----------------|
| PLA DISPERSION |
| BioORMOCER®    |
| PLA DISPERSION |
| BASE PAPER     |

} < 10 g/m<sup>2</sup>

**INN PRESSME**  
Open Innovation Test Bed

Koppolu, R., Nissinen, E., Hämäläinen, R, et al., Specialty Papers Europe | 5-6 September 2023



# THANK YOU

## Questions or other remarks?

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